

# FATTORI DI CORREZIONE PER GAS - UNITA' SI

## CORRECTION FACTORS FOR SIZING VALVES FOR GAS - SI UNITS

Le capacità di scarico delle valvole di sicurezza per gas sono normalmente espresse in Nm<sup>3</sup>/h di aria a 0°C e 1013 mbar. Se il gas da scaricare è diverso dall'aria, la temperatura di esercizio del gas diversa da 0°C e la portata volumica riferita a condizioni normali diverse da 1013 mbar e 0°C, la portata volumica normale del gas dovrà essere moltiplicata per i fattori  $F_g$ ,  $F_s$  ed  $F_t$ ; si otterrà la portata equivalente di aria alle condizioni normali delle tabelle delle capacità di scarico delle valvole di sicurezza. Esempio: Calcolare la portata equivalente di aria a 0°C e 1013 mbar per le seguenti condizioni: gas = azoto - temperatura di processo = 160°C - portata = 8500 m<sup>3</sup>/h riferiti a 1013 mbar e 20°C. Dalla tabella C (tipo di gas):  $F_g = 0,98$ . Dalla tabella D (condizioni di riferimento):  $F_s = 0,932$ . Dalla tabella E (temperatura di processo):  $F_t = 1,26$ .

La portata equivalente sarà:  $Q = 8500 \times 0,98 \times 0,932 \times 1,26 = 9782 \text{ Nm}^3/\text{h}$

The discharge capacities of the safety valves are normally expressed in std.m<sup>3</sup>/h of air at 0°C and 1013 mbar. If the gas to be discharged is different from air, the process temperature is different from 0°C and the volume flowrate is referred to standard conditions different from 0°C and 1013 mbar, the volume flowrate of the gas must be multiplied by the factors  $F_g$ ,  $F_s$  ed  $F_t$ ; the equivalent flowrate of air will be obtained which will correspond to the standard conditions of the discharge capacity tables of the safety valves. Example: Find the equivalent flowrate of air at 0°C and 1013 mbar for the following conditions: gas = nitrogen - process temperature = 160°C - flowrate = 8500 m<sup>3</sup>/h at 1013 mbar and 20°C. From table C (type of gas):  $F_g = 0,98$ . From table D (reference conditions):  $F_s = 0,932$ . From table E (process temperature):  $F_t = 1,26$ .

The equivalent flowrate will be:  $Q = 8500 \times 0,98 \times 0,932 \times 1,26 = 9782 \text{ Nm}^3/\text{h}$

Tabella C  
Table C

Gas o vapore - Gas or vapour	Formula	M kg/kmole	$\rho$ kg/m <sup>3</sup>	V m <sup>3</sup> /kg	R J/kg-K	x	$F_g$
Acetilene - Acetylene	<chem>C2H2</chem>	26,04	1,162	0,861	319,3	1,245	0,99
Acetone - Acetone	<chem>C3H6O</chem>	58,08	2,591	0,386	143,2	1,131	1,53
Acido cloridrico - Hydrochloric acid	<chem>HCl</chem>	36,46	1,627	0,615	228,0	1,397	1,12
Alcool etilico - Ethyl alcohol	<chem>C2H6O</chem>	46,07	2,055	0,487	180,5	1,134	1,36
Alcool metilico - Methyl alcohol	<chem>CH4O</chem>	32,04	1,429	0,700	259,5	1,240	1,10
Ammoniaca - Ammonia	<chem>NH3</chem>	17,03	0,760	1,316	488,2	1,313	0,78
Anidride carbonica - Carbon dioxide	<chem>CO2</chem>	44,01	1,964	0,509	188,9	1,300	1,27
Anidride solforosa - Sulphur dioxide	<chem>SO2</chem>	64,06	2,858	0,350	129,8	1,284	1,53
Argo - Argon	<chem>Ar</chem>	39,95	1,782	0,561	208,1	1,665	1,11



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Gas o vapore - Gas or vapour	Formula	M kg/kmole	$\rho$ kg/m <sup>3</sup>	V m <sup>3</sup> /kg	R J/kg-K	x	F <sub>g</sub>
Aria secca - Dry air	-	(28,96)	1,293	0,773	287,1	1,404	1,00
Azoto - Nitrogen	N <sub>2</sub>	28,01	1,250	0,800	296,8	1,425	0,98
Benzolo - Benzole	C <sub>6</sub> H <sub>6</sub>	78,11	3,485	0,287	106,4	1,127	1,77
Bromoclorodifluorometano - Bromochlorodifluoromethane	CBrClF <sub>2</sub>	165,36	7,378	0,136	50,3	1,08	2,62
Bromotrifluorometano - Bromotrifluoromethane	CBrF <sub>3</sub>	148,91	6,644	0,151	55,8	1,143	2,44
Bromuro di metile - Methyl bromide	CH <sub>3</sub> Br	94,93	4,235	0,236	87,6	1,25	1,88
Butano - Butane	C <sub>4</sub> H <sub>10</sub>	58,12	2,593	0,386	143,1	1,114	1,54
Butilene - Butylene	C <sub>4</sub> H <sub>8</sub>	56,10	2,503	0,400	148,2	1,20	1,47
Cloro - Chlorine	Cl <sub>2</sub>	70,91	3,164	0,316	117,3	1,329	1,59
Cloroetano - Chloroetane	C <sub>2</sub> H <sub>5</sub> Cl	64,51	2,878	0,347	128,9	1,16	1,60
Cloruro di metile - Methyl chloride	CH <sub>3</sub> Cl	50,49	2,253	0,444	164,7	1,27	1,37
Cloruro di vinile - Vinyl chloride	C <sub>2</sub> H <sub>3</sub> Cl	62,50	2,788	0,359	133,0	1,29	1,51
Diclorofluorometano - Dichlorofluoromethane	CHCl <sub>2</sub>	102,92	4,592	0,218	80,8	1,17	2,01
Diclorometano - Dichloromethane	CH <sub>2</sub> Cl <sub>2</sub>	84,93	3,789	0,264	97,9	1,15	1,84
Diclorotetrafluorometano - Dichlorotetrafluoromethane	C <sub>2</sub> Cl <sub>2</sub> F <sub>4</sub>	170,92	7,626	0,131	48,6	1,084	2,66
Diphil (26,5%difenile+ossido di difenile)	-	165,8	3,802	0,263	50,1	1,05	2,65
Elio - Helium	He	4,003	0,179	5,599	2077,0	1,665	0,35
Etano - Ethane	C <sub>2</sub> H <sub>6</sub>	30,07	1,342	0,745	276,5	1,201	1,08
Etere dietilico - Diethyl ether	C <sub>4</sub> H <sub>10</sub> O	74,12	3,307	0,302	112,2	1,085	1,75
Etilene - Ethilene	C <sub>2</sub> H <sub>4</sub>	28,05	1,251	0,799	296,4	1,255	1,02
Freon 12	CF <sub>2</sub> Cl <sub>2</sub>	120,91	5,394	0,185	68,8	1,13	2,21
Freon 22	CHClF <sub>2</sub>	86,47	3,858	0,259	96,2	1,178	1,84
Freon 502	CHClF <sub>2</sub> /C <sub>2</sub> ClF	121,28	5,411	0,185	68,6	1,135	2,21
Gas di città - Town-gas	-	(11,70)	0,522	1,916	710,6	1,369	0,64
Gas di cocheria - Coal gas	-	(11,90)	0,531	1,883	698,7	1,34	0,65
Idrogeno - Hydrogen	H <sub>2</sub>	2,016	0,090	11,118	4124,2	1,415	0,26
Idrogeno solforato - Hydrogen sulphide	H <sub>2</sub> S	34,08	1,520	0,658	244,0	1,324	1,11
Kripto - Krypton	Kr	83,80	3,739	0,267	99,2	1,65	1,61
Metano - Methane	CH <sub>4</sub>	16,04	0,716	1,397	518,3	1,316	0,76
Metilamina - Methylamine	CH <sub>3</sub> N	31,05	1,385	0,722	267,8	1,20	1,09
Neon - Neon	Ne	20,18	0,900	1,111	412,0	1,64	0,79
Ossido di azoto - Nitrogen oxide	NO	30,06	1,341	0,746	276,6	1,38	1,02

Gas o vapore - Gas or vapour	Formula	M kg/kmole	$\rho$ kg/m <sup>3</sup>	V m <sup>3</sup> /kg	R J/kg-K	x	F <sub>g</sub>
Ossido di carbonio - Carbon monoxide	CO	28,01	1,250	0,800	296,8	1,401	0,98
Ossido di diazoto - Dinitrogen oxide	N <sub>2</sub> O	44,01	1,964	0,509	188,9	1,27	1,28
Ossigeno - Oxygen	O <sub>2</sub>	32,00	1,428	0,700	259,8	1,400	1,05
Ozono - Ozone	O <sub>3</sub>	48,00	2,142	0,467	173,2	1,30	1,32
Pentano - Pentane	C <sub>5</sub> H <sub>12</sub>	72,15	3,219	0,311	115,2	1,08	1,73
Propano - Propane	C <sub>3</sub> H <sub>8</sub>	44,10	1,968	0,508	188,5	1,138	1,33
Propilene - Propylene	C <sub>3</sub> H <sub>6</sub>	42,08	1,877	0,533	197,6	1,130	1,30
Solfuro di carbonio - Carbon disulphide	CS <sub>2</sub>	76,13	3,397	0,294	109,2	1,230	1,70
Tetracloruro di carbonio - Carbon tetrachloride	CCl <sub>4</sub>	153,82	6,863	0,146	54,1	1,116	2,50
Toluolo - Toluol	C <sub>7</sub> H <sub>8</sub>	92,14	4,111	0,243	90,2	1,098	1,95
Tricloro fluorometano - Trichloro fluoromethane	CFCl <sub>3</sub>	137,37	6,129	0,163	60,5	1,13	2,35
Triclorotrifluorometano - Trichlorotrifluoromethane	C <sub>2</sub> Cl <sub>3</sub> F <sub>3</sub>	187,38	8,360	0,120	44,4	1,079	2,79
Trifluorometano - Trifluoromethane	CHF <sub>3</sub>	70,01	3,123	0,320	118,8	1,22	1,63
Trifluorocloruro di metano - Methane trifluorochloride	CF <sub>3</sub> Cl	104,46	4,660	0,215	79,6	1,17	2,02
Triossido di zolfo - Sulphur trioxide	SO <sub>3</sub>	80,06	3,572	0,280	103,9	1,208	1,75

**Nota - Note**

M = massa molare - molar mass;  $\rho$  = massa volumica - mass per unit volume (at 0°C and 1013 mbar); V = volume specifico - specific volume; R = costante del gas - gas constant;  $x = c_p/c_v$  a 0°C e 1013 mbar; F<sub>g</sub> = fattore di correzione - correction factor

Tabella D  
Table D

Pref. (ass. - abs)			t <sub>ref</sub>		F <sub>s</sub>
mbar	kg/cm <sup>2</sup>	mmHg	°C	°F	
1013	1,033	760	0	32	1,000
1013	1,033	760	20	68	0,932
1013	1,033	760	15,6	60	0,945
1013	1,033	760	15	59	0,945
980,7	1,000	735,6	20	68	0,902
980,7	1,000	735,6	0	32	0,968
980,7	1,000	735,6	15	59	0,918

Tabella E  
Table E

t°C	F <sub>t</sub>	t°C	F <sub>t</sub>	t°C	F <sub>t</sub>	t°C	F <sub>t</sub>	t°C	F <sub>t</sub>
0	1,00								
5	1,01	65	1,11	150	1,24	270	1,41	390	1,56
10	1,02	70	1,12	160	1,26	280	1,42	400	1,57
15	1,03	75	1,13	170	1,27	290	1,44	410	1,58
20	1,04	80	1,14	180	1,29	300	1,45	420	1,59
25	1,04	85	1,15	190	1,30	310	1,46	430	1,60
30	1,05	90	1,15	200	1,32	320	1,47	440	1,62
35	1,06	95	1,16	210	1,33	330	1,49	450	1,63
40	1,07	100	1,17	220	1,34	340	1,50	460	1,64
45	1,08	110	1,18	230	1,36	350	1,51	470	1,65
50	1,09	120	1,20	240	1,37	360	1,52	480	1,66
55	1,10	130	1,21	250	1,38	370	1,53	490	1,67
60	1,10	140	1,23	260	1,40	380	1,55	500	1,68

# FATTORI DI CORREZIONE RIEPILOGO

## CORRECTION FACTORS

	<b>ARIA</b> PS = 1,293 kg/m <sup>3</sup>	<b>METANO</b> PS = 0,716 kg/m <sup>3</sup>	<b>BUTANO</b> PS = 2,59 kg/m <sup>3</sup>	<b>PROPANO</b> PS = 1,968 kg/m <sup>3</sup>	<b>GPL</b> PS = 2,08 kg/m <sup>3</sup>
<b>ARIA</b> PS = 1,293 kg/m <sup>3</sup>	1	1,34	0,71	0,81	0,79
<b>METANO</b> PS = 0,716 kg/m <sup>3</sup>	0,74	1	0,53	0,60	0,59
<b>BUTANO</b> PS = 2,59 kg/m <sup>3</sup>	1,41	1,90	1	1,15	1,12
<b>PROPANO</b> PS = 1,968 kg/m <sup>3</sup>	1,23	1,66	0,87	1	0,97
<b>GPL</b> PS = 2,08 kg/m <sup>3</sup>	1,27	1,70	0,90	1,03	1



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